

## FDS for advanced

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# Presentation outline

Heat transfer

Fire definition

Using FDS  
outputs

- 1 Heat transfer modeling in FDS
- 2 Fire definition
- 3 Using FDS outputs

# Heat transfer

Heat transfer

Fire definition

Using FDS  
outputs

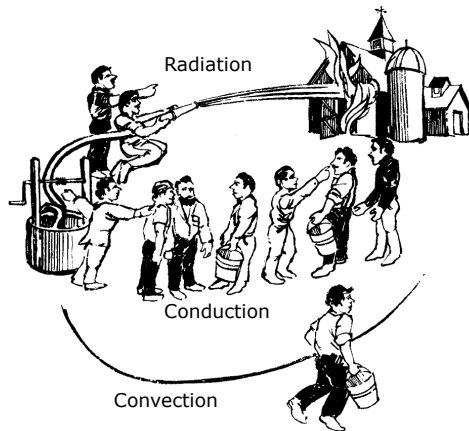


# Heat transfer

Heat transfer

Fire definition

Using FDS  
outputs



Adapted from : A Heat Transfer Textbook, 4th edition John H. Lienhard IV, Professor, University of Houston, John H. Lienhard V, Professor, Massachusetts Institute of Technology, Copyright (c) 2000-2011, John H. Lienhard IV and John H. Lienhard V. All rights reserved.

# Heat conduction basics

- Very weak within the gases
- First propagation mechanism of heat within solids
- Fourier's law :

$$j = -\lambda \nabla T$$

- Heat equation :

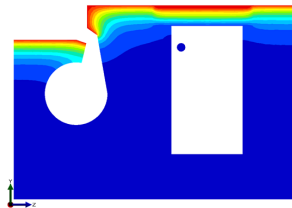
$$\nabla \cdot (\lambda \nabla T) + \dot{q} = \rho C_p \frac{\partial T}{\partial t}$$

- One dimensional form, without heat source and with homogeneous  $\lambda$  :

$$\frac{\partial T}{\partial t} = \alpha \frac{\partial^2 T}{\partial x^2}$$

where

$$\alpha = \frac{\lambda}{\rho C_p} \quad \text{thermal diffusivity}$$



# Heat convection basics

Heat transfer

Fire definition

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- Fluid movement is required
- **Natural convection** : governed by density differences
- **Forced convection** : governed by the forced flow
- Convective flux :

$$\varphi = h(T_{\text{fluid}} - T_{\text{solid}})$$

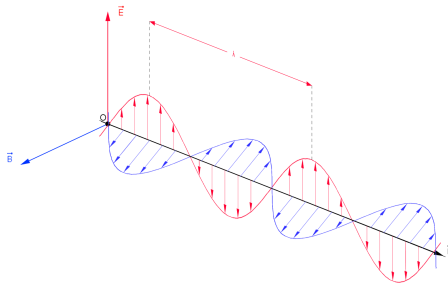
- The convection coefficient  $h$  is strongly linked to the flow characteristics
- FDS formulation :

$$h = \text{MAX} \left[ C|\Delta T|^{1/3}; \frac{k}{L} 0.037 Re^{4/5} Pr^{1/3} \right]$$



# Heat radiation basics

- Electromagnetic wave (Maxwell equations)
  - No material support is required
  - The two main characteristics are wavelength and intensity
- All solids are continuously exchanging radiation
- Radiation is the most important heat transfer mechanism in fires
- **Blackbody**
  - A body which absorbs all incoming radiation
  - Diffuse radiation : no preferred direction



# Blackbody radiant emittance

Heat transfer

Fire definition

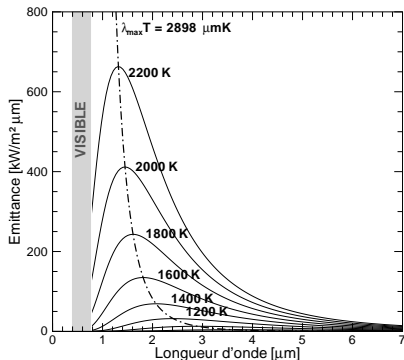
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- **Radiant emittance** : total energy per unit time per unit surface, radiated by an elementary surface
- Blackbody radiant emittance → Planck's function :

$$e_{\lambda,b} = \frac{2hc^2\lambda^{-5}}{\exp(\frac{hc_0}{\lambda kT}) - 1}$$

- Stefan law :

$$e_b = \int e_{\lambda,b} d\lambda = \sigma T^4$$





# Radiative balance

Heat transfer

Fire definition

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- Emissivity of a material
  - Ratio between radiation emitted by the material and the radiation emitted by a blackbody at the same temperature

$$\varepsilon_{\lambda} = \frac{e_{\lambda}}{e_{\lambda,b}} \in [0; 1]$$

- Common assumption :  $\varepsilon_{\lambda}$  does not change with wavelength (gray surfaces)

$$e = \varepsilon \sigma T^4$$

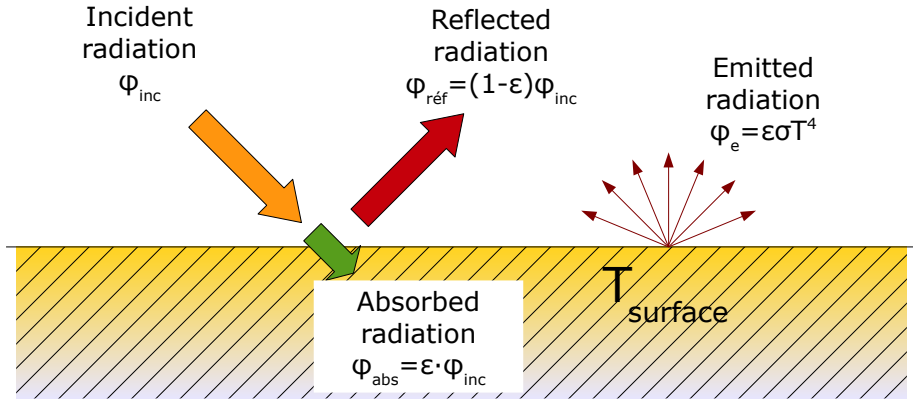
- Kirchoff law : absorbed fraction of incoming radiation = emissivity

# Radiative balance

Heat transfer

Fire definition

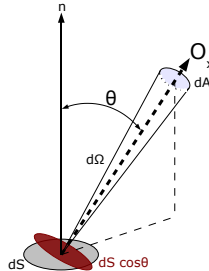
Using FDS  
outputs



## Radiant intensity

- **Radiant intensity**  $I_{O_x}$  emitted by an elementary surface  $dS$  in a direction  $O_x$  = radiant flux emitted in a solid angle  $\Omega$  around  $O_x$  direction, divided by the projection of the source on the direction

$$I_{O_x} = \frac{d\phi_x}{d\Omega dS \cos \theta} \quad (1)$$



# Radiative exchange between solids

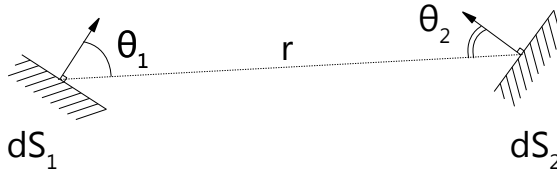
Heat transfer

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- Radiative flux emitted by an elementary surface  $dS_1$  with radiant intensity  $I_1$  to another elementary surface  $dS_2$  :

$$d\phi = I_1 d\Omega dS_1 \cos \theta_1 = I_1 \frac{dS_1 \cos \theta_1 dS_2 \cos \theta_2}{r^2} \quad (2)$$



View factors catalog  
<http://www.thermalradiation.net/indexCat.html>

## Radiant intensity and emittance

- *Diffuse* radiation = radiant intensity is independent of direction  $O_x$  and :

$$\begin{aligned}\frac{1}{dS} \int \phi_x &= \int_{2\pi} L_{O_x} \cos \theta \Omega \\ &= L \int_0^{2\pi} \int_0^{\pi/2} \sin \theta \cos \theta d\theta d\phi \\ &= \pi I\end{aligned}\tag{3}$$

- Link between radiant intensity and emittance  $I = e/\pi$
- Blackbody :  $I_b = \sigma T^4/\pi$
- In FDS, output INTEGRATED INTENSITY is :

$$U = \int I dS \quad \text{At } 20^\circ \text{C} : \quad U = 4\pi I_b = 4\sigma T^4\tag{4}$$

# Interaction between radiation and gases

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- Ambient air : no effect, transparent
- Smoke : participating media
  - **Absorption** → Radiant energy is partially accumulated by electrically asymmetric molecules ( $\text{CO}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{CH}_4$ , etc.) and soot
  - **Diffusion** → Radiant energy is partially deviated by particles

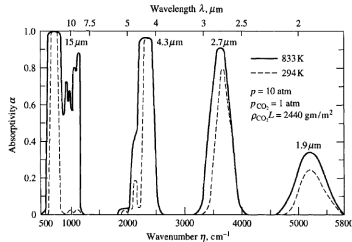
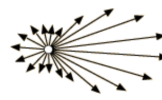


FIGURE 1-15  
Spectral absorptivity of an isothermal mixture of nitrogen and carbon dioxide, from Edwards [16].

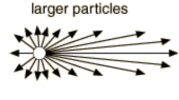
Rayleigh Scattering



Mie Scattering



Mie Scattering,  
larger particles



→ Direction of incident light

# Radiative transfer equation (RTE)

Heat transfer

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outputs

- Radiant intensity variation  $I_\lambda$  along the path  $s$  :

$$\frac{dI_\lambda}{ds} = \underbrace{-\kappa I_\lambda}_{\text{absorption}} + \underbrace{\kappa I_{\lambda,b}}_{\text{emission}} - \underbrace{\sigma I_\lambda + \frac{\sigma}{4\pi} \int_{4\pi} \Phi I_\lambda d\Omega}_{\text{diffusion}}$$

- Radiant attenuation governed by the absorption coefficient  $\kappa$ 
  - Inhomogeneous
  - Linked the the gases composition (mainly soot density)

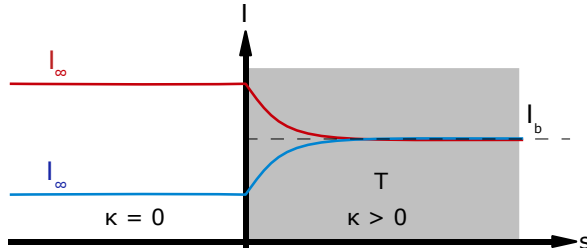
# Radiative transfer equation : a simple example

- Monodimensional case, homogeneous media

$$\frac{dI}{ds} = \kappa(I_b - I) = \kappa\left(\frac{\sigma T^4}{\Pi} - I\right) \quad \text{where} \quad I(0) = I_\infty$$

- Solution

$$I(s) = (I_\infty - I_b) \exp(-\kappa \cdot s) + I_b$$





# Radiative transfer equation in FDS

Heat transfer

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Using FDS  
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- Spectral decomposition

$$I(\mathbf{x}, \mathbf{s}) = \int_0^{\infty} I_{\lambda}(\mathbf{x}, \mathbf{s}) d\lambda \quad (5)$$

- RTE

$$\begin{aligned} \frac{dI_{\lambda}(\mathbf{x}, \mathbf{s})}{d\mathbf{s}} &= \mathbf{s} \cdot \nabla I_{\lambda}(\mathbf{x}, \mathbf{s}) \\ &= -[\kappa(\mathbf{x}, \lambda) + \sigma(\mathbf{x}, \lambda)] I_{\lambda}(\mathbf{x}, \mathbf{s}) + B(\mathbf{x}, \lambda) \\ &\quad + \frac{\sigma(\mathbf{x}, \lambda)}{4\pi} \int_{4\pi} \Phi(\mathbf{s}, \mathbf{s}') I_{\lambda}(\mathbf{x}, \mathbf{s}') \Omega \end{aligned} \quad (6)$$

- Simplification : non diffusive media

$$\mathbf{s} \cdot \nabla I_{\lambda}(\mathbf{x}, \mathbf{s}) = \kappa(\mathbf{x}, \lambda) [I_b(\mathbf{x}) - I_{\lambda}(\mathbf{x}, \mathbf{s})] \quad (7)$$

# Radiative transfer equation in FDS

- Spectral bands decomposition

$$\mathbf{s} \cdot \nabla I_n(\mathbf{x}, \mathbf{s}) = \kappa_n(\mathbf{x}) [I_{b,n}(\mathbf{x}) - I_n(\mathbf{x}, \mathbf{s})] \quad \text{with } n = 1, \dots, N \quad (8)$$

- Total radiative intensity

$$I(\mathbf{x}, \mathbf{s}) = \sum_{n=1}^N I_n(\mathbf{x}, \mathbf{s}) \quad (9)$$

- Radiative flux vector divergence and integrated intensity

$$-\nabla \cdot \mathbf{q}_r = \sum_{n=1}^N \kappa_n(\mathbf{x}) [U_n(\mathbf{x}) - 4\pi L_{b,n}(\mathbf{x})] \quad (10)$$

$$U_n(\mathbf{x}) = \int_{4\pi} I_n(\mathbf{x}, \mathbf{s}) \Omega \quad (11)$$

Absorption coefficients  $\kappa_n$  depend on temperature and gas composition (**RadCal**)

# Radiative transfer equation in FDS

Heat transfer

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- In standard use of FDS : gray gas
  - Radiation is driven by soot
  - Continuous spectrum for soot
  - $\Rightarrow$  one band model

$$\mathbf{s} \cdot \nabla I(\mathbf{x}, \mathbf{s}) = \kappa(\mathbf{x}) \left[ \frac{\sigma T(\mathbf{x})^4}{\pi} - I(\mathbf{x}, \mathbf{s}) \right] \quad (12)$$

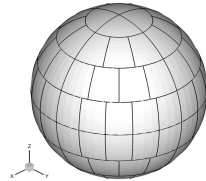
- Then, use the solution to evaluate

$$-\nabla \cdot \mathbf{q}_r = \kappa(\mathbf{x}) \left[ \int_{4\pi} I(\mathbf{x}, \mathbf{s}) \Omega - 4\sigma T(\mathbf{x})^4 \right] \quad (13)$$

# Radiative transfer equation in FDS

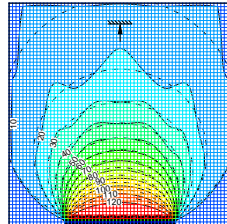
- Angular discretisation
- Special topics
  - Combustion zone

$$\kappa I_b = \max \left( \chi_r \frac{\dot{q}''' }{4\pi}, \kappa \frac{\sigma T(\mathbf{x})^4}{\pi} \right)$$



(14)

- Strongly linked to gas composition ( $\kappa$ ), radiation in  $T^4$ , fluctuations, etc.
- There are some preferential directions



Heat transfer

Fire definition

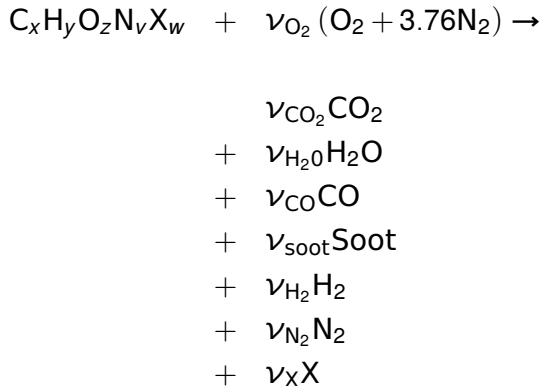
Using FDS  
outputs

# Fire definition



# Combustion modeling

- Global chemical equation



(15)

# Heat of combustion

Heat transfer

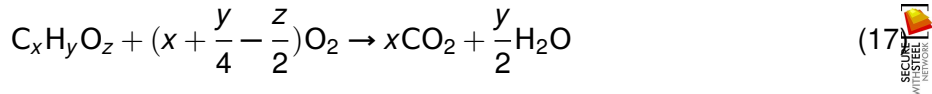
Fire definition

Using FDS  
outputs

- Total energy obtained by combustion

$$E = m_{O_2} \cdot \Delta H_{O_2} \quad (16)$$

- $\Delta H_{O_2} \simeq 13,1 \text{ MJ/kg}$
- Ideal combustion of  $C_xH_yO_z$



$$\Delta H_c = \frac{32(x + y/4 - z/2)}{12x + y + 16z} \Delta H_{O_2} \quad (18)$$



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## Conversion factors

- Link between consumption and production of species

$$m_k = \frac{\nu_k}{\nu_{O_2}} \frac{M_k}{M_{O_2}} m_{O_2} = \frac{\nu_k}{\nu_{O_2}} \frac{M_k}{M_{O_2}} \frac{E}{\Delta H_{O_2}} \quad (19)$$

- Conversion factor

$$\tau_k = \frac{m_k}{E} = \frac{\nu_k M_k}{\Delta H_{O_2} \nu_{O_2} M_{O_2}} \quad (20)$$

- Soot and fuel

$$\tau_{\text{soot}} = \frac{m_{\text{soot}}}{E} = \frac{\nu_{\text{soot}} M_{\text{soot}}}{\Delta H_c M_{\text{fuel}}} = \frac{y_{\text{soot}}}{\Delta H_c} \quad (21)$$



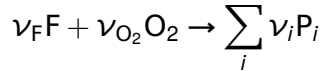
# Mixture fraction

Heat transfer

Fire definition

Using FDS  
outputs

- Combustion equation



- Transport equation

$$\frac{\partial \rho Y_k}{\partial t} + \nabla \cdot (\rho \mathbf{u} Y_k) = \nabla \cdot (\rho D_k \nabla Y_k) + \dot{\omega}_k \quad (22)$$

- Production and consumption are linked by :

$$\dot{\omega}_{O_2} = s \dot{\omega}_F \quad \text{et} \quad \dot{\omega}_P = -(1 + s) \dot{\omega}_F \quad (23)$$

$$\text{with } s = \frac{\nu_{O_2} M_{O_2}}{\nu_F M_F}$$

# Mixture fraction

- Mixture fraction

$$Z = \frac{sY_F - (Y_{O_2} - Y_{O_2}^\infty)}{sY_F^I + Y_{O_2}^\infty} \quad (24)$$

- Infinitely fast chemistry means

$$Z_{st} = \frac{Y_{O_2}^\infty}{sY_F^I + Y_{O_2}^\infty} \quad (25)$$

- Transport equation for mixture fraction

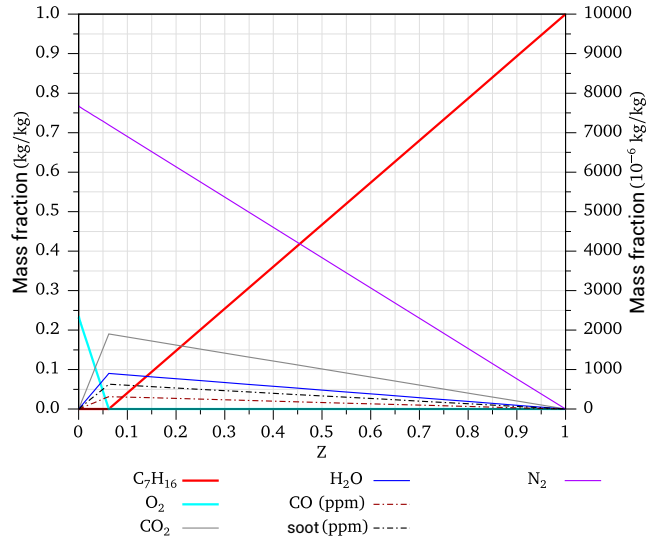
$$\frac{\partial \rho Z}{\partial t} + \nabla \cdot (\rho \mathbf{u} Z) = \nabla \cdot (\rho D \nabla Z) \quad (26)$$

## Example : heptane

Heat transfer

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Using FDS  
outputs



## Simple extinction model

Heat transfer

Fire definition

Using FDS  
outputs

- Oxygen combustion :  $Q = m \cdot Y_{O_2} \frac{\Delta H_f}{r_{O_2}}$
- Adiabatic flame temperature :  $T_f = T_0 + \frac{Q}{m \cdot \overline{C_p}}$
- Hence :  $Y_{O_2} = \frac{\overline{C_p}(T_f - T_0)}{\Delta H_f / r_{O_2}}$
- With typical values of  $\overline{C_p}$ ,  $T_f$  et  $\Delta H_f / r_{O_2}$  :

$$Y_{O_2, \text{lim}} = \frac{1.2(1\,700 - T_0)}{13\,100}$$

(27)   
SECURE  
WITH STEEL  
NETWORK

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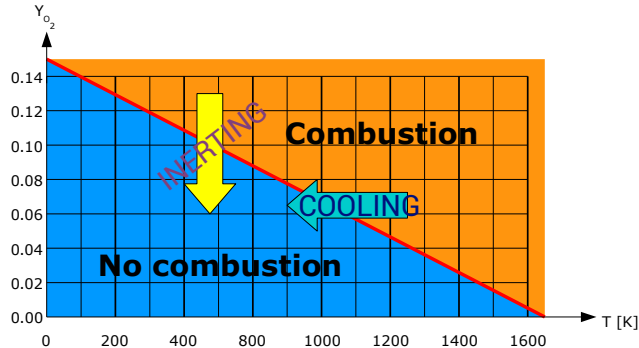
# Simple extinction model

Heat transfer

Fire definition

Using FDS  
outputs

$$Y_{O_2, \text{lim}} = \frac{1.2(1\,700 - T_0)}{13\,100}$$



# Using FDS outputs

Heat transfer

Fire definition

Using FDS  
outputs

# Radiative balance at solid surface

Heat transfer

Fire definition

Using FDS  
outputs

- Radiative flux emitted

$$\varphi_e^{\text{rad}} = \varepsilon \sigma T^4 \quad (28)$$

- Received radiative flux

$$\varphi_r^{\text{rad}} = \varepsilon \varphi_{\text{incident}}^{\text{rad}} \quad (29)$$

- Net radiative flux

$$\varphi_{\text{net}}^{\text{rad}} = \varphi_r^{\text{rad}} - \varphi_e^{\text{rad}} = \varepsilon (\varphi_{\text{incident}}^{\text{rad}} - \sigma T^4) \quad (30)$$

# Thermal balance at solid surface

- Convective balance

$$\phi_{\text{net}}^{\text{conv}} = h(T_g - T) \quad (31)$$

- Thermal balance

$$\phi_{\text{net}} = \phi_{\text{net}}^{\text{rad}} + \phi_{\text{net}}^{\text{conv}} = \varepsilon(\phi_{\text{incident}}^{\text{rad}} - \sigma T^4) + h(T_g - T) \quad (32)$$

- Written as

$$\phi_{\text{net}} = \underbrace{(\varepsilon \phi_{\text{incident}}^{\text{rad}} + h T_g)}_{(a)} - \underbrace{(\varepsilon \sigma T^4 + h T)}_{(b)}$$

- (a) is the thermal environment calculated by FDS
- (b) is the thermal response of the surface

Extracting (a) from  
FDS calculation?



# FDS usual outputs

Heat transfer

Fire definition

Using FDS  
outputs

- CONVECTIVE\_HEAT\_FLUX :  $\varphi_{\text{net}}^{\text{conv}} = h(T_g - T)$
- RADIATIVE\_HEAT\_FLUX :  $\varphi_{\text{net}}^{\text{rad}} = \varepsilon(\varphi_{\text{incident}}^{\text{rad}} - \sigma T^4)$
- NET\_HEAT\_FLUX :  $\varphi_{\text{net}} = \varphi_{\text{net}}^{\text{conv}} + \varphi_{\text{net}}^{\text{rad}}$
- INCIDENT\_HEAT\_FLUX :  $\frac{\varphi_{\text{net}}^{\text{rad}}}{\varepsilon} + \sigma T^4 + h(T_g - T)$
- GAUGE\_HEAT\_FLUX :  $\varphi_{\text{net}}^0 = \varphi_{\text{incident}}^{\text{rad}} - \sigma T_{\infty}^4 + h(T_g - T_{\infty})$
- RADIOMETER :  $\varphi^{\text{RD}} = \frac{\varphi_{\text{net}}^{\text{rad}}}{\varepsilon} + \sigma(T^4 - T_{\infty}^4)$
- None of them is really adapted to transfer information to FEM model

# Adiabatic Surface Temperature

Heat transfer

Fire definition

Using FDS  
outputs

- Root of

$$\varphi_{\text{net}} = \varepsilon \sigma (T_{\text{AST}}^4 - T^4) + h(T_{\text{AST}} - T) \quad (34)$$

- Temperature of an adiabatic surface ( $\varphi_{\text{net}} = 0$ )
- Temperature seen by the surface (used exactly as the ISO curve for input in FEM heat conduction calculation)
- (a) quantity can be calculated as

$$\varepsilon \varphi_{\text{incident}}^{\text{rad}} + h \cdot T_g = \varepsilon \sigma T_{\text{AST}}^4 + h T_{\text{AST}} \quad (35)$$

- $T_{\text{AST}}$  is the best way of transferring heat fluxes on surfaces normal to the main axis

Heat transfer

Fire definition

Using FDS  
outputs

