

FDS for advanced

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Presentation outline

Heat transfer

Fire definition

Using FDS
outputs

- 1 Heat transfer modeling in FDS
- 2 Fire definition
- 3 Using FDS outputs

Heat transfer

Heat transfer

Fire definition

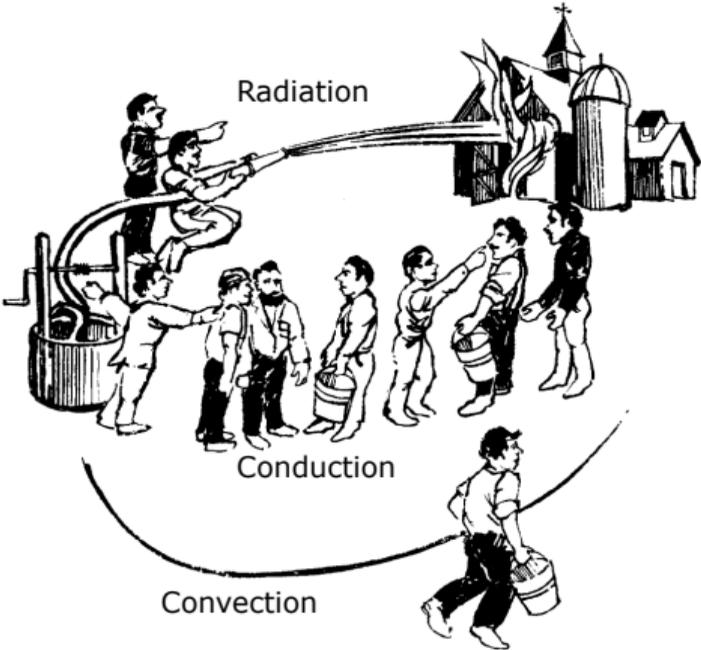
Using FDS
outputs

Heat transfer

Heat transfer

Fire definition

Using FDS
outputs



Adapted from : A Heat Transfer Textbook, 4th edition John H. Lienhard IV, Professor, University of Houston, John H. Lienhard V, Professor, Massachusetts Institute of Technology, Copyright (c) 2000-2011, John H. Lienhard IV and John H. Lienhard V. All rights reserved.

Heat conduction basics

- Very weak within the gases
- First propagation mechanism of heat within solids
- Fourier's law :

$$j = -\lambda \nabla T$$

- Heat equation :

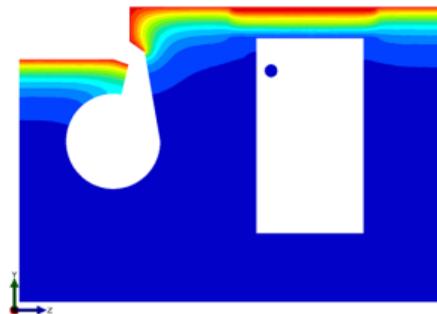
$$\nabla \cdot (\lambda \nabla T) + \dot{q} = \rho C_p \frac{\partial T}{\partial t}$$

- One dimensional form, without heat source and with homogeneous λ :

$$\frac{\partial T}{\partial t} = \alpha \frac{\partial^2 T}{\partial x^2}$$

where

$$\alpha = \frac{\lambda}{\rho C_p} \quad \text{thermal diffusivity}$$



Heat convection basics

- Fluid movement is required
- **Natural convection** : governed by density differences
- **Forced convection** : governed by the forced flow
- Convective flux :

$$\varphi = h(T_{\text{fluid}} - T_{\text{solid}})$$

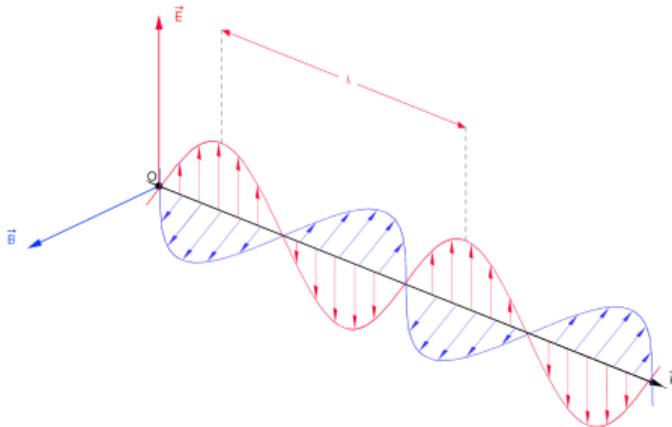
- The convection coefficient h is strongly linked to the flow characteristics
- FDS formulation :

$$h = \text{MAX} \left[C|\Delta T|^{1/3}; \frac{k}{L} 0.037 \text{Re}^{4/5} \text{Pr}^{1/3} \right]$$



Heat radiation basics

- Electromagnetic wave (Maxwell equations)
 - No material support is required
 - The two main characteristics are wavelength and intensity
- All solids are continuously exchanging radiation
- Radiation is the most important heat transfer mechanism in fires
- **Blackbody**
 - A body which absorbs all incoming radiation
 - Diffuse radiation : no preferred direction



Blackbody radiant emittance

Heat transfer

Fire definition

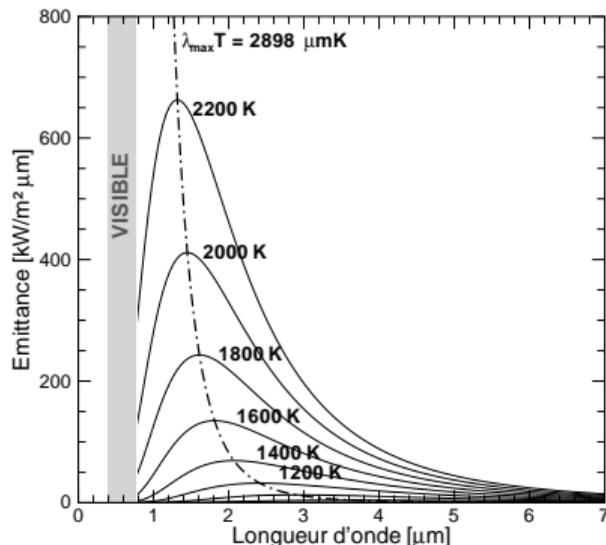
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- **Radiant emittance** : total energy per unit time per unit surface, radiated by an elementary surface
- Blackbody radiant emittance → Planck's function :

$$e_{\lambda,b} = \frac{2hc_0^2\lambda^{-5}}{\exp\left(\frac{hc_0}{\lambda kT}\right) - 1}$$

- Stefan law :

$$e_b = \int e_{\lambda,b} d\lambda = \sigma T^4$$



Radiative balance

Heat transfer

Fire definition

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- Emissivity of a material
 - Ratio between radiation emitted by the material and the radiation emitted by a blackbody at the same temperature

$$\varepsilon_{\lambda} = \frac{e_{\lambda}}{e_{\lambda,b}} \in [0; 1]$$

- Common assumption : ε_{λ} does not change with wavelength (gray surfaces)

$$e = \varepsilon \sigma T^4$$

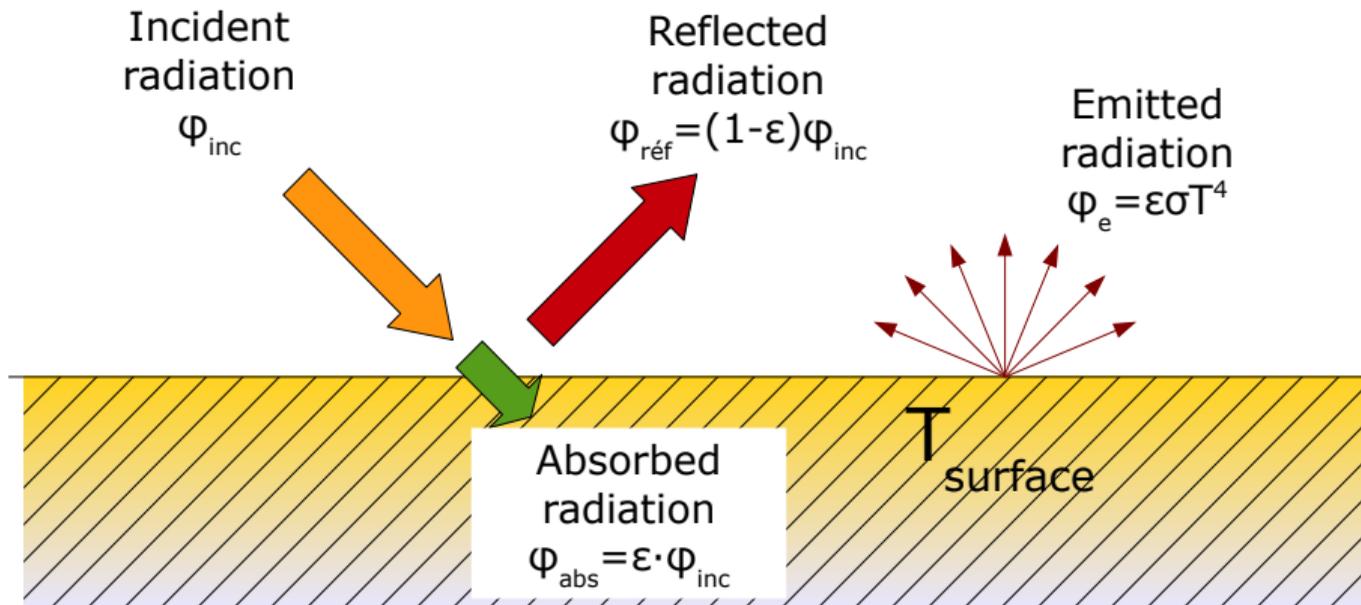
- Kirchoff law : absorbed fraction of incoming radiation = emissivity

Radiative balance

Heat transfer

Fire definition

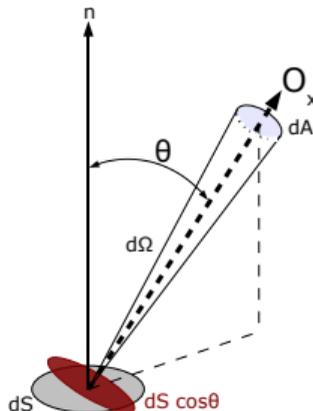
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Radiant intensity

- **Radiant intensity** I_{O_x} emitted by an elementary surface dS in a direction O_x = radiant flux emitted in a solid angle Ω around O_x direction, divided by the projection of the source on the direction

$$I_{O_x} = \frac{d\phi_x}{d\Omega dS \cos \theta} \quad (1)$$



Radiative exchange between solids

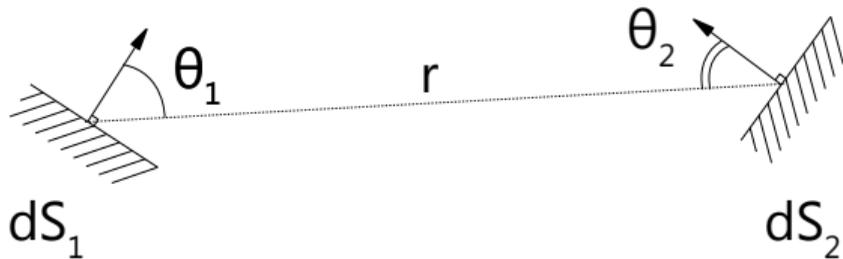
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- Radiative flux emitted by an elementary surface dS_1 with radiant intensity I_1 to another elementary surface dS_2 :

$$d\phi = I_1 d\Omega dS_1 \cos \theta_1 = I_1 \frac{dS_1 \cos \theta_1 dS_2 \cos \theta_2}{r^2} \quad (2)$$



View factors catalog
<http://www.thermalradiation.net/indexCat.html>

Radiant intensity and emittance

- *Diffuse* radiation = radiant intensity is independent of direction O_x and :

$$\begin{aligned}\frac{1}{dS} \int \varphi_x &= \int_{2\pi} L_{O_x} \cos \theta \Omega \\ &= L \int_0^{2\pi} \int_0^{\pi/2} \sin \theta \cos \theta \theta d\phi \\ &= \pi l\end{aligned}\tag{3}$$

- Link between radiant intensity and emittance $l = e / \pi$
- Blackbody : $l_b = \sigma T^4 / \pi$
- In FDS, output INTEGRATED INTENSITY is :

$$U = \int l dS \quad \text{At } 20 \text{ } ^\circ\text{C} : \quad U = 4\pi l_b = 4\sigma T^4\tag{4}$$

Interaction between radiation and gases

Heat transfer

Fire definition

Using FDS
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- Ambient air : no effect, transparent
- Smoke : participating media
 - **Absorption** → Radiant energy is partially accumulated by electrically asymmetric molecules (CO_2 , H_2O , CH_4 , etc.) and soot
 - **Diffusion** → Radiant energy is partially deviated by particles

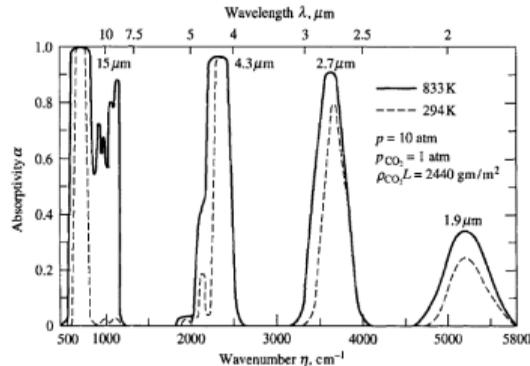
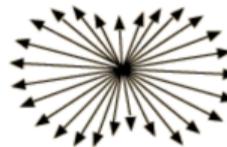
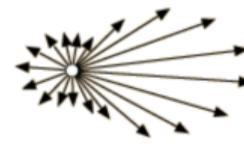


FIGURE 1-15 Spectral absorptivity of an isothermal mixture of nitrogen and carbon dioxide, from Edwards [16].

Rayleigh Scattering



Mie Scattering



Mie Scattering,
larger particles



→ Direction of incident light

Radiative transfer equation (RTE)

- Radiant intensity variation I_λ along the path s :

$$\frac{dI_\lambda}{ds} = \underbrace{-\kappa I_\lambda}_{\text{absorption}} + \underbrace{\kappa I_{\lambda,b}}_{\text{emission}} - \underbrace{\sigma I_\lambda + \frac{\sigma}{4\pi} \int_{4\pi} \Phi I_\lambda d\Omega}_{\text{diffusion}}$$

- Radiant attenuation governed by the absorption coefficient κ
 - Inhomogeneous
 - Linked to the gases composition (mainly soot density)

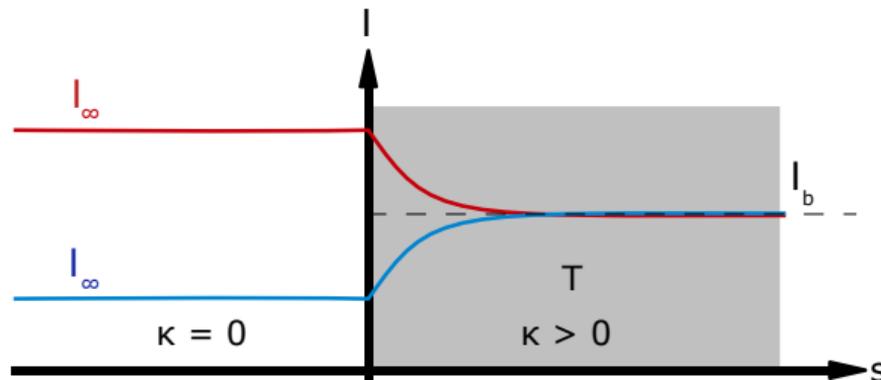
Radiative transfer equation : a simple example

- Monodimensional case, homogeneous media

$$\frac{dI}{ds} = \kappa(I_b - I) = \kappa\left(\frac{\sigma T^4}{\Pi} - I\right) \quad \text{where} \quad I(0) = I_\infty$$

- Solution

$$I(s) = (I_\infty - I_b) \exp(-\kappa \cdot s) + I_b$$



Radiative transfer equation in FDS

Heat transfer

Fire definition

Using FDS
outputs

- Spectral decomposition

$$I(\mathbf{x}, \mathbf{s}) = \int_0^{\infty} I_{\lambda}(\mathbf{x}, \mathbf{s}) d\lambda \quad (5)$$

- RTE

$$\begin{aligned} \frac{dI_{\lambda}(\mathbf{x}, \mathbf{s})}{d\mathbf{s}} &= \mathbf{s} \cdot \nabla I_{\lambda}(\mathbf{x}, \mathbf{s}) \\ &= -[\kappa(\mathbf{x}, \lambda) + \sigma(\mathbf{x}, \lambda)] I_{\lambda}(\mathbf{x}, \mathbf{s}) + B(\mathbf{x}, \lambda) \\ &\quad + \frac{\sigma(\mathbf{x}, \lambda)}{4\pi} \int_{4\pi} \Phi(\mathbf{s}, \mathbf{s}') I_{\lambda}(\mathbf{x}, \mathbf{s}') \Omega \end{aligned} \quad (6)$$

- Simplification : non diffusive media

$$\mathbf{s} \cdot \nabla I_{\lambda}(\mathbf{x}, \mathbf{s}) = \kappa(\mathbf{x}, \lambda) [I_b(\mathbf{x}) - I_{\lambda}(\mathbf{x}, \mathbf{s})] \quad (7)$$

Radiative transfer equation in FDS

- Spectral bands decomposition

$$\mathbf{s} \cdot \nabla I_n(\mathbf{x}, \mathbf{s}) = \kappa_n(\mathbf{x}) [I_{b,n}(\mathbf{x}) - I_n(\mathbf{x}, \mathbf{s})] \quad \text{with } n = 1, \dots, N \quad (8)$$

- Total radiative intensity

$$I(\mathbf{x}, \mathbf{s}) = \sum_{n=1}^N I_n(\mathbf{x}, \mathbf{s}) \quad (9)$$

- Radiative flux vector divergence and integrated intensity

$$-\nabla \cdot \mathbf{q}_r = \sum_{n=1}^N \kappa_n(\mathbf{x}) [U_n(\mathbf{x}) - 4\pi L_{b,n}(\mathbf{x})] \quad (10)$$

$$U_n(\mathbf{x}) = \int_{4\pi} I_n(\mathbf{x}, \mathbf{s}) \Omega \quad (11)$$

Absorption coefficients κ_n depend on temperature and gas composition (RadCal)

Radiative transfer equation in FDS

- In standard use of FDS : gray gas
 - Radiation is driven by soot
 - Continuous spectrum for soot
 - \Rightarrow one band model

$$\mathbf{s} \cdot \nabla I(\mathbf{x}, \mathbf{s}) = \kappa(\mathbf{x}) \left[\frac{\sigma T(\mathbf{x})^4}{\pi} - I(\mathbf{x}, \mathbf{s}) \right] \quad (12)$$

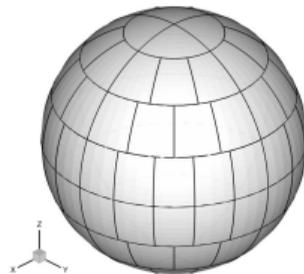
- Then, use the solution to evaluate

$$-\nabla \cdot \mathbf{q}_r = \kappa(\mathbf{x}) \left[\int_{4\pi} I(\mathbf{x}, \mathbf{s}) \Omega - 4\sigma T(\mathbf{x})^4 \right] \quad (13)$$

Radiative transfer equation in FDS

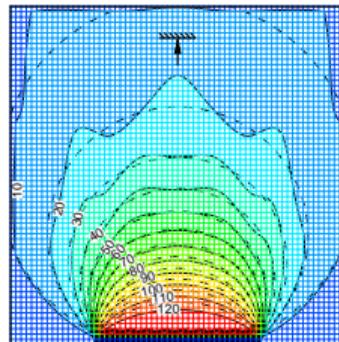
- Angular discretisation
- Special topics
 - Combustion zone

$$\kappa l_b = \max \left(\chi_r \frac{\dot{q}'''}{4\pi}, \kappa \frac{\sigma T(\mathbf{x})^4}{\pi} \right)$$



(14)

- Strongly linked to gas composition (κ), radiation in T^4 , fluctuations, etc.
- There are some preferential directions



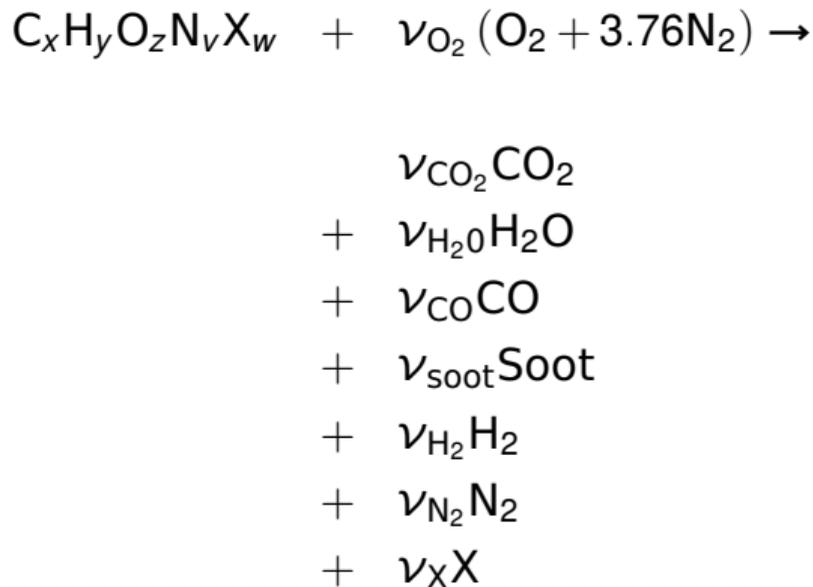
Fire definition

Heat transfer

Fire definition

Using FDS
outputs

- Global chemical equation

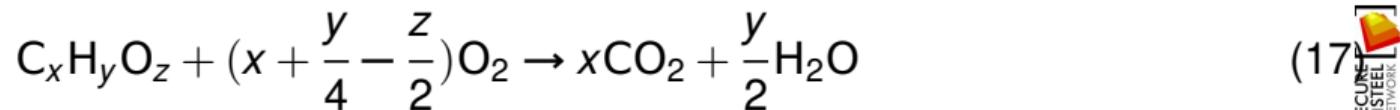


Heat of combustion

- Total energy obtained by combustion

$$E = m_{O_2} \cdot \Delta H_{O_2} \quad (16)$$

- $\Delta H_{O_2} \simeq 13,1$ MJ/kg
- Ideal combustion of $C_xH_yO_z$



$$\Delta H_c = \frac{32(x + y/4 - z/2)}{12x + y + 16z} \Delta H_{O_2} \quad (18)$$

Conversion factors

- Link between consumption and production of species

$$m_k = \frac{\nu_k}{\nu_{O_2}} \frac{M_k}{M_{O_2}} m_{O_2} = \frac{\nu_k}{\nu_{O_2}} \frac{M_k}{M_{O_2}} \frac{E}{\Delta H_{O_2}} \quad (19)$$

- Conversion factor

$$\tau_k = \frac{m_k}{E} = \frac{\nu_k M_k}{\Delta H_{O_2} \nu_{O_2} M_{O_2}} \quad (20)$$

- Soot and fuel

$$\tau_{\text{soot}} = \frac{m_{\text{soot}}}{E} = \frac{\nu_{\text{soot}} M_{\text{soot}}}{\Delta H_c M_{\text{fuel}}} = \frac{y_{\text{soot}}}{\Delta H_c} \quad (21)$$

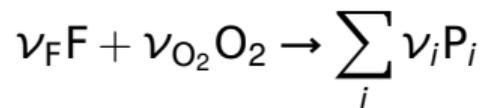
Mixture fraction

Heat transfer

Fire definition

Using FDS
outputs

- Combustion equation



- Transport equation

$$\frac{\partial \rho Y_k}{\partial t} + \nabla \cdot (\rho \mathbf{u} Y_k) = \nabla \cdot (\rho D_k \nabla Y_k) + \dot{\omega}_k \quad (22)$$

- Production and consumption are linked by :

$$\dot{\omega}_{O_2} = s \dot{\omega}_F \quad \text{et} \quad \dot{\omega}_P = -(1 + s) \dot{\omega}_F \quad (23)$$

$$\text{with } s = \frac{\nu_{O_2} M_{O_2}}{\nu_F M_F}$$

Mixture fraction

- Mixture fraction

$$Z = \frac{sY_F - (Y_{O_2} - Y_{O_2}^\infty)}{sY_F' + Y_{O_2}^\infty} \quad (24)$$

- Infinitely fast chemistry means

$$Z_{st} = \frac{Y_{O_2}^\infty}{sY_F' + Y_{O_2}^\infty} \quad (25)$$

- Transport equation for mixture fraction

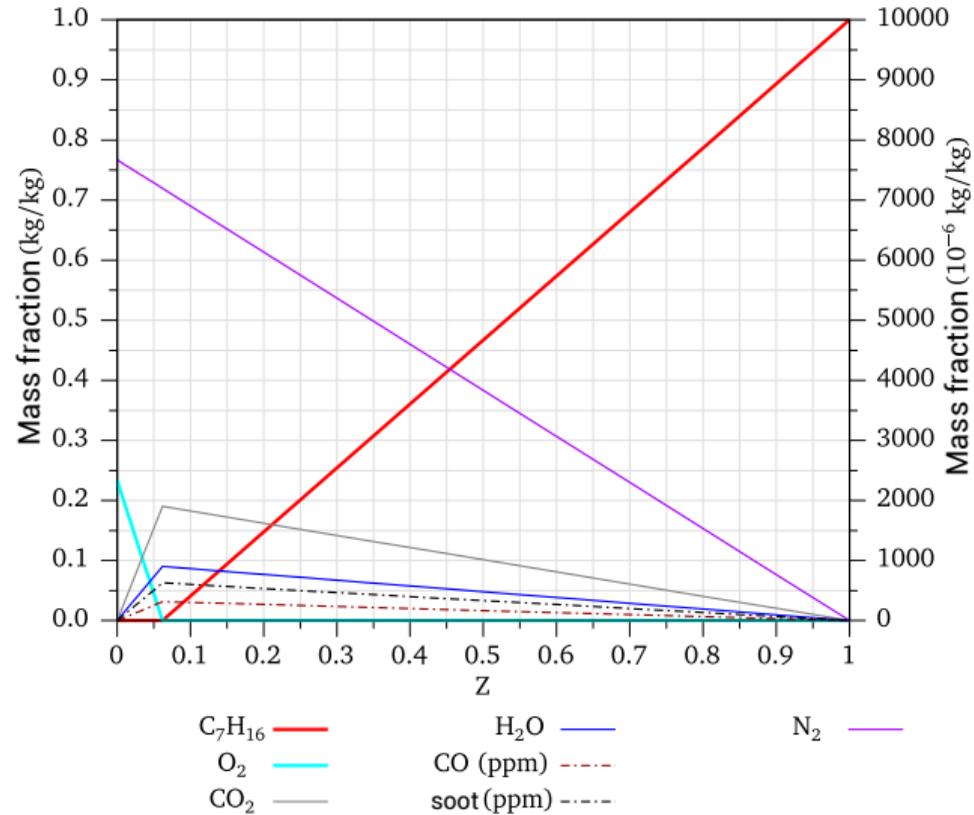
$$\frac{\partial \rho Z}{\partial t} + \nabla \cdot (\rho \mathbf{u} Z) = \nabla \cdot (\rho D \nabla Z) \quad (26)$$

Example : heptane

Heat transfer

Fire definition

Using FDS
outputs



Simple extinction model

Heat transfer

Fire definition

Using FDS
outputs

- Oxygen combustion : $Q = m \cdot Y_{O_2} \frac{\Delta H_f}{r_{O_2}}$
- Adiabatic flame temperature : $T_f = T_0 + \frac{Q}{m \cdot \overline{C_p}}$
- Hence : $Y_{O_2} = \frac{\overline{C_p}(T_f - T_0)}{\Delta H_f / r_{O_2}}$
- With typical values of $\overline{C_p}$, T_f et $\Delta H_f / r_{O_2}$:

$$Y_{O_2, \text{lim}} = \frac{1.2(1\ 700 - T_0)}{13\ 100}$$

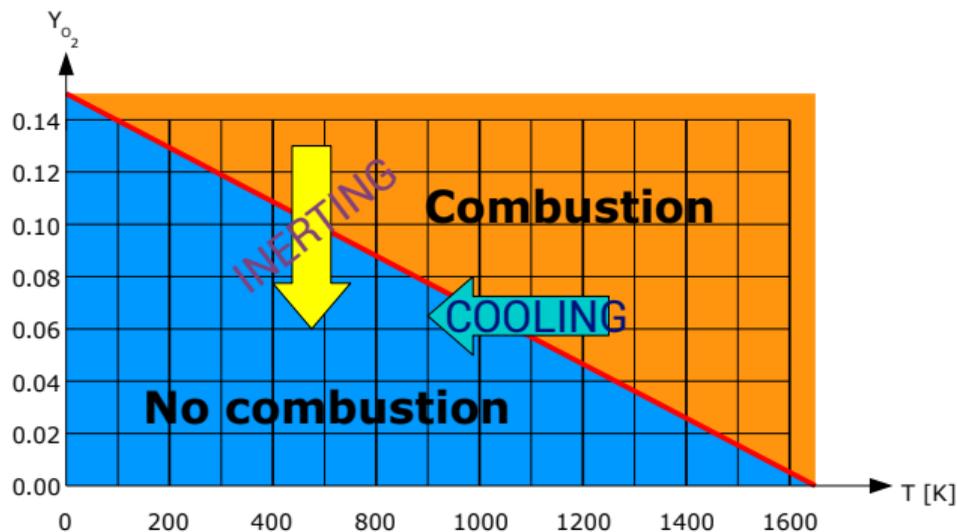
Simple extinction model

Heat transfer

Fire definition

Using FDS
outputs

$$Y_{O_2, \text{lim}} = \frac{1.2(1700 - T_0)}{13100}$$



Using FDS outputs

Heat transfer

Fire definition

Using FDS
outputs

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Radiative balance at solid surface

Heat transfer

Fire definition

Using FDS
outputs

- Radiative flux emitted

$$\varphi_e^{\text{rad}} = \varepsilon \sigma T^4 \quad (28)$$

- Received radiative flux

$$\varphi_r^{\text{rad}} = \varepsilon \varphi_{\text{incident}}^{\text{rad}} \quad (29)$$

- Net radiative flux

$$\varphi_{\text{net}}^{\text{rad}} = \varphi_r^{\text{rad}} - \varphi_e^{\text{rad}} = \varepsilon (\varphi_{\text{incident}}^{\text{rad}} - \sigma T^4) \quad (30)$$

Thermal balance at solid surface

- Convective balance

$$\varphi_{\text{net}}^{\text{conv}} = h(T_g - T) \quad (31)$$

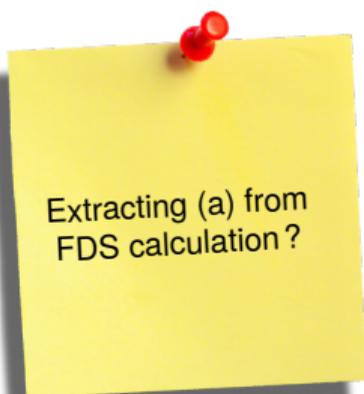
- Thermal balance

$$\varphi_{\text{net}} = \varphi_{\text{net}}^{\text{rad}} + \varphi_{\text{net}}^{\text{conv}} = \varepsilon(\varphi_{\text{incident}}^{\text{rad}} - \sigma T^4) + h(T_g - T) \quad (32)$$

- Written as

$$\varphi_{\text{net}} = \underbrace{(\varepsilon\varphi_{\text{incident}}^{\text{rad}} + hT_g)}_{(a)} - \underbrace{(\varepsilon\sigma T^4 + hT)}_{(b)}$$

- (a) is the thermal environment calculated by FDS
- (b) is the thermal response of the surface



Extracting (a) from
FDS calculation?

FDS usual outputs

Heat transfer

Fire definition

Using FDS
outputs

- CONVECTIVE_HEAT_FLUX : $\varphi_{\text{net}}^{\text{conv}} = h(T_g - T)$
- RADIATIVE_HEAT_FLUX : $\varphi_{\text{net}}^{\text{rad}} = \varepsilon(\varphi_{\text{incident}}^{\text{rad}} - \sigma T^4)$
- NET_HEAT_FLUX : $\varphi_{\text{net}} = \varphi_{\text{net}}^{\text{conv}} + \varphi_{\text{net}}^{\text{rad}}$
- INCIDENT_HEAT_FLUX : $\frac{\varphi_{\text{net}}^{\text{rad}}}{\varepsilon} + \sigma T^4 + h(T_g - T)$
- GAUGE_HEAT_FLUX : $\varphi_{\text{net}}^0 = \varphi_{\text{incident}}^{\text{rad}} - \sigma T_{\infty}^4 + h(T_g - T_{\infty})$
- RADIOMETER : $\varphi^{\text{RD}} = \frac{\varphi_{\text{net}}^{\text{rad}}}{\varepsilon} + \sigma(T^4 - T_{\infty}^4)$
- None of them is really adapted to transfer information to FEM model

Adiabatic Surface Temperature

Heat transfer

Fire definition

Using FDS
outputs

- Root of

$$\varphi_{\text{net}} = \varepsilon\sigma(T_{\text{AST}}^4 - T^4) + h(T_{\text{AST}} - T) \quad (34)$$

- Temperature of an adiabatic surface ($\varphi_{\text{net}} = 0$)
- Temperature seen by the surface (used exactly as the ISO curve for input in FEM heat conduction calculation)
- (a) quantity can be calculated as

$$\varepsilon\varphi_{\text{incident}}^{\text{rad}} + h \cdot T_g = \varepsilon\sigma T_{\text{AST}}^4 + hT_{\text{AST}} \quad (35)$$

- T_{AST} is the best way of transferring heat fluxes on surfaces normal to the main axis

Heat transfer

Fire definition

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